



Chemistry 112, General Chemistry Laboratory B

Spring 2022 Syllabus

The following syllabus applies to all of the *in-person* lab sections: Chem 112-001 to 112-012.

Lab Location: Flanner Hall 305

Students can only attend the section in which they are enrolled in LOCUS. Be mindful of day/time and location. All days and times are central standard time (CST).

Pre/Co-requisite: Chem 111 or 105

Prerequisite: Math Placement Test or Math 117

Laboratory Coordinator: Dr. Katrina Binaku

Office Hours in ZOOM: [Tuesdays 11am-12pm](#), [Fridays 2:30-3pm](#), and by a scheduled appointment.

Office Location: 104 Flanner Hall

Office Phone: 773-508-8715

Email: kbinaku@luc.edu

Laboratory Coordinator: Dr. Andrew Basner

Office Hours: Tuesdays 12-2 pm via zoom and in person by scheduled appointment.

Office Hours Zoom Info: <https://luc.zoom.us/my/abasner>

Office Location: 428 Flanner Hall

Office Phone: 773-508-3135

Email: abasner@luc.edu

Laboratory Coordinator: Dr. Y. Corey Lin

Office Hours: Thursday 1-3 pm via zoom and in person by scheduled appointment.

Office Hours Zoom Info: <https://luc.zoom.us/j/89469519169>

Office Location: 104 Flanner Hall

Office Phone: 773-508-2598

Email: ylin21@luc.edu

Teaching Assistants (TAs) will be assisting all of the laboratory coordinators during this course. Specific TAs and TA information will be listed in Sakai, along with the Primary Lab Coordinator for your section.

Welcome to Chem 112. We are looking forward to working with you this semester. Read the entire syllabus to understand the expectations of this course.

COURSE DESCRIPTION

This 2nd semester general chemistry lab course continues emphasis of lab experiments & data collection, data interpretation/analysis, and scientific writing. Lab topics are related to some topics/theory covered in the lecture course (Chem 102). The course introduces students to intermolecular forces, matter and phase changes, chromatography, chemical kinetics, chemical equilibrium, acids & bases, continued use of net ionic equations, and redox reactions. Continued exposure and enhancement of scientific writing skills is achieved through formal lab reports & keeping a lab notebook. Dimensional analysis and significant figures also play a continued role in the success of students in the course. These lists are not exhaustive but note the course highlights.

Goals of course:

- 1) Extend laboratory skill knowledge built on Chem 111 basics.
- 2) Apply qualitative and quantitative techniques to collect experimental data and appropriately interpret experimental results.
- 3) Improve scientific writing skills and documentation of laboratory data and analysis.
- 4) Learn to work safety in the lab and gain an understanding of chemical and laboratory safety practices.

Outcomes:

- 1) Demonstrate intermolecular forces, chemical kinetics, equilibrium, and acid – base theory through successful lab completion, data analysis & interpretation in notebook entries, online quizzes, and formal lab report writing.
- 2) Connect calculations to course concepts through successful lab notebook experiment calculations, data interpretation, and online quizzes.
- 3) Demonstrate enhanced scientific writing skills through experimental lab notebook and scientific lab reports.

REQUIRED ITEMS

- 1) Chem 112 Laboratory Packet (provided as a PDF in Sakai). Printed manuals stored in the lab are provided during synchronous sessions. Feel free to print out a personal copy of the manual if you prefer to.
- 2) Composition style notebook (not spiral bound & cannot have tear-out perforations). Line ruled. The notebook used for Chem 111 can be used again, so long as there is adequate space.
- 3) Safety goggles (we provide to you). These must be type G, H or K goggles and must meet or exceed ANSI Z87.1 requirements. Safety glasses do not meet our requirements and are not allowed.
- 4) A face mask. See Spring 2022 Masking Requirement below.
- 5) Long-sleeve Laboratory Coat (white is preferred coat color). You must purchase this.
- 6) Appropriate clothing and footwear. See below for details*

- 7) Scientific calculator for most course work. Cell phones are not calculators.
- 8) A non-erasable pen is required for all written work. No white-out is allowed. No pencils allowed.
- 9) Sakai access via the internet to watch pre-lab video content, post-lab content, online content, lab simulations, and complete all Sakai work.
- 10) Cam Scanner app OR a scanner machine. Cam scanner is a free app that converts a phone picture to a PDF file. You will take pictures of the Composition lab notebook pages and convert them to a PDF file in order to submit the notebook pages to Sakai for grading. This app works on android and iPhone. A scanner machine can be used as well.

***Appropriate clothing must be worn that minimizes potential chemical contact with your skin. Shoes that adequately cover the entire foot are required. Sandals, open-toes shoes, perforated shoes, open-backed shoes are not acceptable. No skin should be exposed on your feet or legs, so clothing that covers and protects your body from the waist down (including your ankles) must be worn. Lab coat required.**

SPRING 2022 MASKING REQUIREMENT

It is Departmental policy that, even in the event the University relaxes its universal requirement for indoor mask-wearing during the Spring 2022 semester, it will remain a principle of this class-section that, out of respect for the health of housemates and others in regular contact with members of our community, in this class we properly wear masks at all times (e.g. over nose and mouth).

GENERAL POLICIES

- Attendance is mandatory. All in-person written & Sakai work, as well as TA observations, serves as the basis for earning points and showing progress. Written work will be graded with an emphasis on correct significant digits, consistent results (do data & observations match conclusions), appropriateness/correctness of analysis, and thoroughness in responses. Following directions of reporting calculated answers are taken into account in grades too. There are no online supplements to in-person lab experiments; students have to be in attendance, in person to complete lab work.
- All work must be completed in non-erasable pen. This includes the Composition notebook, any worksheets/handouts and homework assignments. Work not completed in pen or containing “white-out” is subject to a point deduction and is not eligible for any regrade requests.
- The Composition notebook is to contain all laboratory experiment information required according to the lab notebook guidelines. It is the major component of your course grade. Losing this Composition notebook may result in failure of the course, as one cannot be graded on work that does not exist if the notebook is

lost. Follow the guidelines given on how to appropriately set-up a laboratory notebook.

- Aspects of lab work must be done in the Composition notebook only. (No loose-leaf paper or other notebooks). Your TA will evaluate/grade your notebook. This functions as a way for TAs to grade along the way and to make sure you are on the right track with lab results. If there are any discrepancies in recorded grades, proof of having earned a specific grade on a particular lab is the presence of that graded work in your notebook.
- Each student is assigned a drawer with glassware and equipment. At the beginning of the semester & semester's end, the drawer contents will be checked for completeness. The drawer is shared with other students over the course of a week. Therefore, it is essential that you clean the equipment used after an experiment is done. Drawers may be checked sporadically. If glassware is broken, the student is responsible for requesting a replacement item; there is no penalty for broken glassware.
- Homework can never be submitted via email. No exceptions.
- In general, Sakai work cannot be made up. (**See Late Work Policies section**).
- If absence occurs, in-lab experiment work must be made up during specific dates and times outlined by the Lab Coordinator. Students cannot choose when to makeup work; documentation of illness, reasons for absences will be requested.
- Safety and Clean-up points will be earned on the basis of safe/professional conduct in the lab. A safe lab environment is essential. Unsafe actions will definitely result in grade degradation. The following is a partial list of ways you can lose safety/clean-up points:
 - Coming late to class, after the pre-lab lecture has started will result in deduction of safety points.
 - Not dressing appropriately for lab. Proper footwear/clothing are required.
 - Not bringing goggles to lab/not wearing your goggles consistently in lab can result in expulsion from the lab. Safety glasses do not meet our safety requirements.
 - Not properly wearing a face mask.
 - Not bringing a lab coat to lab. Not wearing the lab coat properly [buttoned] during lab.
 - Not keeping your equipment drawer or lab space in good condition (i.e. dirty glassware/bench).
 - Engaging in horseplay/actions that may endanger you, your classmates, TA, or Lab Coordinator.
 - Not adhering to Disposal Instructions indicated in each lab handout.
 - The lab-pro equipment used is breakable and requires special care. You and your partner will be assigned a box to use, and if the equipment is found to have been handled negligently, points will be deducted from both your safety points and your lab score for the both of you.

An action, even if not herein, deemed unsafe by TA or Lab Coordinator will result in safety point deductions.

Failure to adhere to lab safety rules can result in expulsion from the lab session and/or course with no opportunity for make-up of the work. Safety must be taken very seriously.

ATTENDANCE/PARTICIPATION

Attendance is mandatory. You are required to come to class on your assigned “in-person days” and can only attend the lab section in which you are officially enrolled in LOCUS. There is a point value associated with the work accomplished in each class, and you cannot earn points for classes that you do not attend. **There are limited opportunities to make up missed labs!** A student may attend class during a session listed as asynchronous to complete the make-up work in most instances. There are eight lab experiments and students are expected to complete all of them. If a student is absent for both the lab and make up lab, a zero (0) is recorded in the gradebook for the experimental work missed. Sample data is given so the student can complete the homework questions pertaining to the lab missed.

Students are not allowed to make up a lab experiment in another section of Chemistry 112. If the university is open, you are expected to attend class and be on time. Points are deducted for those who arrive late to lab. If you arrive after the conclusion of the pre-lab lecture, you may not be allowed to perform the lab. Being sent home for improper clothing/footwear also counts as an absence and no makeup work is allowed.

Review the schedule at the end of the syllabus and consider the negative impact that missing a hands-on laboratory session will have on your educational experience. It is in your best interest to register for a section that does not conflict with other obligations. Students should not enroll in a lab section that they cannot fully attend. Missing 2 of the labs, which is nearly 25% of the lab work, is significant and unacceptable and will result in academic failure. This is also true as well for not turning in the assigned typed formal lab report(s).

Loyola University Absence Policy for Students in Co-Curricular Activities (including ROTC):

Students missing classes while representing Loyola University Chicago in an official capacity (e.g. intercollegiate athletics, debate team, model government organization) shall be allowed by the faculty member of record to make up any assignments and to receive notes or other written information distributed in the missed classes.

Students should discuss with faculty the potential consequences of missing lectures and the ways in which they can be remedied. Students must provide their instructors with proper documentation (develop standard form on web) describing the reason for and date of the absence.

This documentation must be signed by an appropriate faculty or staff member, and it must be provided as far in advance of the absence as possible. It is the responsibility of the student to make up any assignments.

<https://www.luc.edu/athleteadvising/attendance.shtml>

Students participating in co-curricular activities must make information concerning time conflicts with University sponsored events available to the Laboratory Coordinator within the first two weeks of the semester. The Laboratory Coordinator reserves the right to contact the Athletics Department confirming time conflicts and regarding concerns. Students missing classes while representing Loyola University Chicago in an official capacity (e.g. intercollegiate athletics, debate team, model government organization) will need to discuss their needs with the Laboratory Coordinator. Sakai work cannot be made up in any circumstances, no exceptions. Laboratory work cannot be made up either; you cannot attend another lab section. These types of absences are handled on a case-by-case basis with remedy.

Accommodations for religious reasons:

Students missing a lab experiment due to observing religious holidays must alert the Lab Coordinator no later than two weeks after the start of the semester to request a special accommodation. This is handled on a case by case basis. The Lab Coordinator reserves the right to contact Campus Ministry, which keeps information on a plethora of religions and holidays.

Students must discuss with the Lab Coordinator the consequences of missing laboratory and the ways [if any] they can be remedied, while also providing the Laboratory Coordinator with proper documentation describing the reason and date of the absence. The document must be signed by an appropriate Faculty/Staff member, and it must be provided as far in advance of the absence as possible. It is a student's responsibility to proactively ask what will be missed due to absence.

HEALTH, SAFETY, AND WELL-BEING ON CAMPUS

Please be familiar with and adhere to all guidelines posted on the *On-Campus Guidelines in Classroom Scenarios*: [Campus Info & Resources](#) and [Return to Campus Guidelines site](#).

COURSE REPEAT RULE

Effective as of the Fall 2017 semester, students are allowed only THREE attempts to pass Chemistry courses with a C- or better grade. The three attempts include withdrawals (W). After the second attempt, the student must secure approval for a third attempt. Students must come to the Chemistry Department, fill out a permission to

register form or print it from the Department of Chemistry & Biochemistry website: <https://luc.edu/chemistry/forms/> and personally meet and obtain a signature from either the Undergraduate Program Director, Assistant Chairperson, or Chairperson in Chemistry. A copy of this form is then taken to your Academic Advisor in Sullivan to secure final permission for the attempt.

PASS/FAIL CONVERSION DEADLINES AND AUDIT POLICY

A student may request to convert a course into or out of the “Pass/No-Pass” or “Audit” status only within the first two weeks of the semester. For the Spring 2022 semester, students are able to convert a class to “Pass/No-Pass” or “Audit” through Monday, January 31st. Students must submit a request for Pass/No-Pass or Audit to their Academic Advisor.

PRIVACY STATEMENT

Assuring privacy among faculty and students engaged in online and face-to-face instructional activities helps promote open and robust conversations and mitigates concerns that comments made within the context of the class will be shared beyond the classroom. As such, any recordings of instructional activities occurring in online or face-to-face classes may be used solely for internal class purposes by the faculty member and students registered for the course, and only during the period in which the course is offered. Students will be informed of such recordings by a statement in the syllabus for the course in which they will be recorded. Instructors who wish to make subsequent use of recordings that include student activity may do so only with informed written consent of the students involved or if all student activity is removed from the recording. Recordings including student activity that have been initiated by the instructor may be retained by the instructor only for individual use.

ROLE OF TEACHING ASSISTANTS

In lab sessions, your primary interaction could be with a Teaching Assistant (TA). The function of a TA is to help you safely get good data and provide individual help when necessary. The role of the Laboratory Coordinator is more behind the scenes: plan the curriculum, prepare handouts and PowerPoints, and train the TAs so the lab experience is educational, fair, and effectively run for students enrolled in all of the sections. Know that the Laboratory Coordinator will be in lab too, but may step outside of the lab from time to time to handle appropriate curriculum work. The Laboratory Coordinator is available to you during and outside of the laboratory hours if there are any questions or concerns that the TAs cannot handle appropriately. The Laboratory Coordinator has final authority in all matters relating to the course. TAs will keep office hours, which will be posted on Sakai. Utilize both your Lab Coordinator and TA if you need assistance. *If

at any point during the semester you want to talk to a Laboratory Coordinator regarding your TA, please do. The TA should enhance your educational experience. If this is not the case, talk to a Lab Coordinator so they know.

TUTORING

The Tutoring Center offers free tutoring. Visit the [Tutoring Center Page](#).

GRADING

Reference the grading scale below. There will be no change in the grading scale nor the number of points allotted in this course. It is in your benefit to attend all 8 in person lab experiments to know the content for homework, quiz, and/or formal lab report. A zero (0) is earned for work not completed. If you do not bring your lab notebook to lab, points will be deducted. There is no final exam in this course.

The University uses the +/- grading scale system and that system is implemented in this course. Rounding only applies to the final course grade percentage. Sakai reports a course grade to TWO digits past the decimal (XX.XX%); this final course grade percentage is rounded to the closest integer. For example, an 89.50% or 89.90% (B+) rounds up to a 90% (A-), BUT an 89.30% or 89.45% (B+) rounds to the integer 89% (B+).

Grades are posted on Sakai within one week of completing the work [Excel homework, notebook entries, Sakai pre/post work]. Any grading discrepancies must be resolved the day the graded work is handed back/grade posted in Sakai. Discrepancies in Sakai [grade incorrect, for example] must be resolved no later than one week after reviewing the graded assignment. A student must show proof the work was graded wrong or entered in the Sakai gradebook incorrectly. Grade disputes will not be entertained past 1-week of the grade /graded work being returned to students nor be acknowledged the last day of class. Be mindful of this policy. Efforts are made to ensure that all Chem 112 TAs uniformly grade. On very rare occasions, if it is found there are differences between TA's grading a lab section's averages may be scaled to adjust. When this is necessary, the average lab score mean is adjusted to the average quiz mean.

See the next page for an itemized list for all graded course work.

Grading Scale:

% total	Grade
94 – 100	A
90 – 93	A-
87 – 89	B+
84 – 86	B
80 – 83	B-
77 – 79	C+
74 – 76	C
70 – 73	C-
65 – 69	D+
60 – 64	D
0 – 59	F

Point Breakdown:

Activity	Origin	Points	% of Final Grade
Paper Chromatography	In Lab/Sakai (Assignments)	20	45%
Melting Points of Fatty Acids	In Lab/Sakai (Assignments)	20	
Sherlock Holmes	In Lab/Sakai (Assignments)	20	
Iodine Kinetics	In Lab/Sakai (Assignments)	20	
Bromothymol Blue Equilibrium	In Lab/Sakai (Assignments)	20	
Advanced Acid-Base Titration pKa and Molar Mass of Nicotinic Acid	In Lab/Sakai (Assignments)	20	
Total Lab Notebook Points		120	
Quiz 1 – Melting Points of Fatty Acids	Sakai (Tests & Quizzes)	20	20%
Quiz 2 – Sherlock Holmes	Sakai (Tests & Quizzes)	20	
Quiz 3 – Iodine Kinetics	Sakai (Tests & Quizzes)	20	
Quiz 4 – Bromothymol Blue Equilibrium	Sakai (Tests & Quizzes)	20	
Quiz 5 – Advanced Acid-Base Titration pKa and Molar Mass of Nicotinic Acid	Sakai (Tests & Quizzes)	20	
Total Quiz Points	Lowest Quiz Dropped	80	
Sci-Finder Scholar/Google Scholar Activity	Sakai (Assignments)	20	5%
Total Scholar Points		20	
Formal Laboratory Report 1, Peer Review	Sakai (Assignments)	15	25%
Formal Laboratory Report 1, Final Submission	Sakai (Assignments)	85	
Formal Laboratory Report 2, Final Submission	Sakai (Assignments)	100	
Total Formal Laboratory Report Points		200	
Clean Up, Safety, Lab Prep	In Lab	10/lab	5%
Total Clean Up, Safety, Lab Prep		80	

Grade if Absent:

A zero (0) is recorded for work not completed, absent or not. The Lab Coordinator also has the right to fail the student if two or more absences occur during the semester or if either typed formal lab report is not turned in.

If you miss a lab, contact your primary Laboratory Coordinator immediately. Lab Coordinator has a right to request documentation to confirm the reason for absence. Request information on whether the in-person lab can be made-up. Request sample data for the lab experiment missed and. The sample data is similar to what you may have obtained in lab and may help you study for any upcoming quiz. You will be responsible for understanding the missed material, and **normal deadlines always apply for completing homework on Sakai.**

Late Work Policies:

QUIZZES: If not completed on time, a 0 is the final grade. Quizzes cannot be accessed after the due date. Answers and feedback are released when the quiz closes.

NOTEBOOK ENTRIES: If not completed on time, a 1-week grade period is allotted to turn the work in late [2pt penalty for lateness applied to grade]. After 1-week, if the notebook is not turned in a 0 is the final grade.

SCHOLAR SEARCH WORK: If not completed on time, a 1-week grade period is allotted to turn the work in late [2pt penalty for lateness applied to grade]. After 1-week, if the work is not turned in a 0 is the final grade.

FORMAL LAB REPORT 1: If the first draft is not completed on time, a 24-hr grace period is allotted because there is a peer review involved. If draft is not turned in, student forfeits 15pts allotted to peer review [since a peer's review cannot be done if a student doesn't turn in a draft of their own typed formal lab report]. For the final draft of the report, a 1-week grade period is allotted to turn the work in late [2pt penalty per day for lateness applied to grade]. After 1-week, if the final draft of the typed formal lab report is not turned in a 0 is the final grade.

FORMAL LAB REPORT 2: If not completed on time, a 1-week grade period is allotted to turn the work in late [2pt penalty per day for lateness applied to grade]. After 1-week, if the lab report is not turned in a 0 is the final grade.

As a student, by enrolling in this course you agree to follow and abide by all syllabus policies & regulations, due dates, and understand grades will suffer if work is not turned in on time or if work is not turned in at all.

EDUCATIONAL GOAL

In this general chemistry laboratory course, my purpose as the Lab Coordinator is to provide a hands-on introduction to experimental methods of scientific investigation in Chemistry. The fundamental models of chemistry discussed in lecture will provide the basis for understanding the experimental laboratory work. Each lab will provide a practical opportunity for students to gain competence with the basic techniques of lab work and the practical experience necessary to understand its significance. It is my wish that this laboratory experience will encourage students who are seeking intellectual challenges along with an understanding of the chemical principles in the laboratory. After all, Chemistry is all around us in our everyday lives!

Conducting experiments and collecting data to test the validity of theories and models requires a different set of skills than those required for success in the lecture part of a general chemistry course. During a laboratory activity, each student's hands, mind, eyes, as well as other senses are focused on the task at hand. Success in the lab involves skills in making perceptive qualitative observations and accurate quantitative measurements.

With each laboratory experiment, relevant questions are posed, and along with TAs, the Lab Coordinator assists each student to execute a laboratory approach which will yield reliable data related to these questions. Each student is required to obtain data and to depend upon this data when answers to these questions are drafted. All labs are structured enough so that the student should not feel lost or confused, but not so structured that a student will find it unnecessary to think for themselves.

REGARDING SAKAI AND TECHNICAL DIFFICULTIES

It is *strongly encouraged* that all required submissions to Sakai as well as writing lab reports, opening course/data/experiment files, be done on a reliable wired internet connection [not wireless], that of which the University itself provides in the Information Commons and various computer labs on the Lake Shore Campus and Downtown.

Under NO circumstances will excuses of "technical difficulties" be accepted as this syllabus is stating all students should use a wired internet University computer [not wireless internet] to submit work in Sakai, write lab reports, open course/data/experiment files. This list is not exhaustive and it should be noted that any activities this course may require a computer or internet connection for should be completed using University computers with wired internet connection. Use of home internet [wired or wireless], University wireless, or public wireless is at your, the student's, own risk. It is not prohibited but as Instructor has stated in this syllabus, Instructor is not responsible for technical difficulties of non-University devices [cell phone, tablet, home/work/public wireless internet or computer]. Do not submit items in Sakai using a cell phone or a tablet device as these do not count as reliable internet connection tools."

ACADEMIC INTEGRITY

The standard of academic integrity and personal honesty delineated in the College of Arts & Sciences Statement on Academic Integrity is expected of every student and will be enforced. Details can be found at http://www.luc.edu/cas/faculty_resources.shtml Cheating can take many forms in lab, but the most common forms are copying data and answers to analysis questions, sharing files for homework, or completing Sakai work with another person. The data and analysis as well as the homework submitted for grading must be your own. If it is not, no credit will be awarded for the entire lab, nor will make-ups be granted. Findings of dishonest academic behavior are reported to the Chair of the Chemistry Department and to the Dean's Office, and are entered into an individual's record. Copied answer/report will result in penalty for all students involved.

DISABILITY ACCOMMODATIONS

If you have a documented disability and wish to discuss academic accommodations, talk to your primary Laboratory Coordinator by the second week of lab. The Coordinator of Student Accessibility Center (SAC), formerly referred to as SSWD, is located in the Sullivan Center and must be contacted independently.

Necessary accommodations will be made for students with disabilities who procure a SAC letter. However, to receive any accommodations self-disclosure, proper documentation, and registration with the SAC office at Loyola University Chicago is required. Accommodations cannot be made until the Laboratory Coordinator receives proper documentation. Furthermore, accommodations are not retro-active and begin only once appropriate documentation has been received by the Laboratory Coordinator in a timely manner. Recognize that the course time scheduled in LOCUS is fixed. No extra time on wet chemistry is given to a student with an SAC letter; it is not possible and the SAC office has been made aware of this.

Only those accommodations that are specifically listed in the formal SAC letter will be provided. If an accommodation letter suggests the Testing Center be utilized to take an exam, it is the student's responsibility to schedule the testing time in the center. Note there are no exams in this course. If flexible attendance and/or extended deadlines are a listed accommodation, the student must meet with the Lab Coordinator and agree to specific terms by filling out the extra form with the Lab Coordinator regarding the accommodations and what can/cannot be offered per the nature of the course. Review the [SAC Policies and Procedures](#).

SMART EVALS

Feedback on the course is important so that a Lab Coordinator can gain insight into how to improve the course, the teaching style, and so the department can learn how best to shape the curriculum for future semesters. Towards the end of the semester, students will receive an email from the Office of Institutional Effectiveness with a reminder to

provide feedback on the Chem 112 course the student is enrolled in. This office will send you constant reminders during the open period of feedback until the evaluation has been completed.

The evaluation is 100% anonymous. When the results are released, no one will be able to tell which student provided the individual feedback. The feedback is not released until after the semester is over, therefore any feedback given will not impact student grades.

LOST AND FOUND

Any items mistakenly left in lab will be taken to the Chemistry Department office, 125 Flanner Hall, and can be identified and claimed there. **Please put your name on your Composition notebook, lab manual, calculators, lab goggles, lab coat [tag], and other personal items.**

SAFETY IN THE LABORATORY

Laboratory safety is everyone's responsibility. By registering for and participating in this course you agree to abide by all of the safety precautions, information, and rules provided to you herein as well as in or outside of the laboratory. Failure to follow these rules constitutes grounds for withdrawing the offending student from the lab session and or course at any time.

The Laboratory Coordinator, TA, and University take safety in the laboratory very seriously. Make sure to always listen to information regarding extra safety precautions when applicable. The rules of safety listed on the following page are reviewed during the first day of the laboratory course. Practice safe laboratory conduct during the entire semester and beyond. This list is not exhaustive and it is the student's responsibility to understand the proper, safe conduct when working in a laboratory. Students cannot complete experiments in the course unless the safety lecture and safety form are completed.

By using common sense and following all of the safety rules provided, it is unlikely that you or your classmates will be involved in or injured in a mishap in the laboratory. While it is very important that you do your part to prevent an accident from occurring, it is just as important to know what to do if someone is injured.

There are several key safety features of a laboratory that will be pointed out during the first day of class.

Preventing an accident or injury from occur is the ideal case scenario, which is why proactively being safe in the laboratory is desired. We live in the real-world and therefore have to be reactive in case of a lab incident.

Although not a requirement, it can be very helpful if a Laboratory Coordinator knows if a student has a condition that could possibly render an unsafe lab situation (allergies to latex, heart condition, seizure risk, etc.). Do feel free to discuss any concerns you may have regarding health conditions and laboratory work.

Your commitment to safety [including the following rules] is very important:

1. To always be on time to lab. Coming in late violates safety. Pre-lab lecture starts on time and missing any of its content is unsafe.
2. To wear a face mask, approved safety goggles¹ and a [buttoned] long-sleeve laboratory coat at all times in the laboratory. Safety glasses are NOT allowed under any circumstances.
3. Non-latex, nitrile glove are optional but *highly* encouraged, especially when working with acids and bases or solvents. Do not wear gloves in the hallway or anywhere outside of lab.
4. To know both the location of and how to use eye washes.
5. Not to wear contacts in the laboratory. Eyeglasses are recommended.
6. To wear appropriate clothing that minimizes potential chemical contact with your skin. A lab coat is required, as are shoes that adequately cover the entire foot. Sandals, open-toe shoes, perforated shoes, open-backed shoes are not acceptable. No skin should be exposed on your feet, ankles, or legs, so clothing that covers and protects your body from the waist down (including ankles) should be worn. You must be dressed appropriately to do experiments.
7. To know both the location of and how to use the safety showers.
8. To know both the location of and how to use the fire extinguishers.
9. To know the proper clean-up and disposal procedure for broken glass.
10. Not to perform unauthorized and unknown experiments, nor work in the lab alone.
11. Not to take chemicals or equipment out of the laboratory.
12. Not to engage in horseplay or any clowning around that may endanger you or other students.
13. Not to eat, drink, chew gum, or smoke anything in the laboratory at any time. No headsets, cell phones, or any other audio devices.
14. Cell phones cannot be used as calculators.
15. To pull long hair back, keeping it away from chemicals and open flame.
16. To keep your lab space clean and tidy. This includes locking your lab locker when done.
17. To ask your Instructor or TA when in doubt about procedures.

18. Inform your Instructor of any health condition you have that might affect your performance or safety in the laboratory.

This list is not exhaustive. The Lab Coordinator and/or Teaching Assistants reserve the right to make a judgement call on an activity they deem unsafe taking place in the laboratory. Safety is a priority and students who do not follow the rules can be removed from the course, and if necessary Campus Safety will be called.

If you have any questions regarding the content of this syllabus, including the safety information provided, you are encouraged to discuss all questions/concerns with the Laboratory Coordinator.

The information provided on the following page are some basic reactive procedures to difference scenarios that have occurred in the laboratory.

FIRST AID BASICS

Minor Cuts: Clean the wound, remove foreign material. Band-Aids are available. Two Band-Aid rule:

If you bleed through one Band-Aid, another should be applied over the first Band-Aid. If you bleed through two Band-Aids in a few minutes, you will be escorted to Health Services). Additionally, if there is any possibility of broken glass in a cut, you will be escorted to the Wellness Center.

Minor Burns from Fire: Immerse affected area in ice water.

Chemicals in Eyes: Immediately flush eyes with water at the eye wash. Continue with flush for at least 10 minutes. Hold the affected eye(s) open to do this properly.

Chemicals on Skin: Rinse affected area with water immediately at the sink or safety shower. If clothing is affected, remove clothes before rinsing! Continue to rinse for at least 10 minutes.

Critical Injuries may include: glass in his/her eye(s), serious cuts, severe chemical burns, severe fire burns, seizures. **Immediately call for help using either the lab phone (security number is taped to phone handle) or the emergency phone in the hallway directly outside the laboratory.** Anyone with chemicals or foreign objects in his/her eye(s) will be escorted to the Wellness Center or to the hospital.

FIRE HAZARDS

The primary heat source in this laboratory is the Bunsen burner, which is fueled by natural gas. A lit Bunsen burner is a small, controllable fire, but the heat generated by the burner fire can be quite hazardous in certain circumstances. It can serve as an ignition source for other combustible materials in the lab, such as paper (lab handouts, paper towels, filter paper, etc.), plastics (wash bottle), flammable liquids (acetone, ethanol). A burner fire can also ignite clothing and hair. Proper operation of a burner

and the absence of combustible materials in the proximity of the burner will significantly reduce the risk of a fire.

Keep chords and paper products away from laboratory hotplates. Always make sure hot plates are off & un-plugged before leaving the lab. Avoid spilling chemicals on hot plates.

Each lab is equipped with several fire extinguishers, fire blanket, and safety showers, which should be used in a fire emergency.

In a case of a fire:

Remain calm; alert the instructor and your immediate neighbors.

Personal safety, yours and others in the labs, is always the top priority.

A small fire in a small container can be suffocated by covering it with a watch glass or inverted beaker.

With a somewhat larger fire, decide whether or not you think you can control it with a fire extinguisher.

Use of a Fire Extinguisher:

Located by the doors in both labs; a back-up fire extinguisher is located at the west end of the floor.

Maintain an escape position; i.e. stay between the fire and the doorway.

Break the plastic ring, pull out the metal ring, release the hose from the bracket, direct the hose at the base of the flames, and press the lever down. PASS (pull, aim, squeeze, sweep).

Note: Fire extinguishers are heavy and not particularly easy to direct. These are multi-purpose, dry chemical extinguishers, safe for anything we use in lab.

The Laboratory Coordinators reserve the right to revise this syllabus in order to correct any unintentional mistakes and/or to change the labs or lab directions for the class if necessary. Students will be notified if any changes have been made.

Tentative Chem 112 Order of Lab Experiments

There are no groups this semester. ALL 48 students are expected to attend lab every week that is listed as Synchronous. The tentative schedule is shown below. Be very mindful that **there are limited make-up opportunities**; do not be absent for an in-person lab unless there is an emergency or valid medical reason.

Labs 1 – 8 are IN-PERSON. There are no online substitutes for lab work!

See next several pages for detailed weekly lab schedule.

Tentative Chem 112 Order of Lab Experiments

Week & Class Dates	Activity Planned
Week 1, Jan 18 th – Jan 21 st	<p>Asynchronous Introduction, syllabus, safety – Panopto lecture</p> <p>Due Next Class Period:</p> <ol style="list-style-type: none"> 1. Read Syllabus 2. Buy required equipment 3. Prepare for course
Week 2, Jan 25 th – Jan 28 th	<p>Asynchronous Google Scholar/Scifinder Scholar Search</p> <p>Due Next Class Period:</p> <ol style="list-style-type: none"> 1. Explore Sakai and the Sakai tools in the course 2. Submit Google Scholar/Scifinder Scholar activity via Sakai Assignments 3. Watch lecture on Paper Chromatography in Sakai (Panopto) 4. Read PDF lab manual for Paper Chromatography experiment (Sakai Resources) 5. Write necessary info in lab notebook, will be checked at start of class.
Week 3, Feb 1 st – Feb 4 th	<p>Synchronous Lab 1 – Paper chromatography, complete and collect classroom data</p> <p>Due Next Class Period:</p> <ol style="list-style-type: none"> 1. Submit lab 1 notebook pages to Sakai (Assignments) 2. Watch lecture on Melting Point of Fatty Acids in Sakai (Panopto) 3. Read PDF lab manual for Melting Point of Fatty Acids experiment (Sakai Resources) 4. Write necessary info in lab notebook, will be checked at start of class.

Week & Class Dates	Activity Planned
Week 4, Feb 8 th – Feb 11 th	<p>Synchronous Lab 2 – Melting point of fatty acids</p> <p>Due Next Class Period:</p> <ol style="list-style-type: none"> 1. Submit lab 2 notebook pages to Sakai (Assignments) 2. Take the quiz on Lab 2 in Sakai (Tests and Quizzes) 3. Watch lecture on Holmes in Sakai (Panopto) 4. Read PDF lab manual for Holmes experiment (Sakai Resources) 5. Write necessary info in lab notebook, will be checked at start of class.
Week 5, Feb 15 th – Feb 18 th	<p>Synchronous Lab 3 – Sherlock Holmes</p> <p>Due Next Class Period:</p> <ol style="list-style-type: none"> 1. Submit lab 3 notebook pages to Sakai (Assignments) 2. Take the quiz on Lab 3 in Sakai (Tests and Quizzes) 3. Watch lecture on Iodine Kinetics in Sakai (Panopto) 4. Read PDF lab manual for Iodine Kinetics experiment (Sakai Resources) 5. Write necessary info in lab notebook, will be checked at start of class.
Week 6, Feb 22 nd – Feb 25 th	<p>Synchronous Lab 4 – Iodine Kinetics</p> <p>Due Next Class Period:</p> <ol style="list-style-type: none"> 1. Submit Lab 4 notebook pages to Sakai (Assignments) 2. Take the quiz on Lab 4 in Sakai (Tests and Quizzes) 3. Watch lecture on Crystal Violet Kinetics in Sakai (Panopto) 4. Read PDF lab manual for Crystal Violet Kinetics experiment (Sakai Resources) 5. Write necessary info in lab notebook, will be checked at start of class.

Week & Class Dates	Activity Planned
Week 7, Mar 1 st – Mar 4 th	<p>Synchronous Lab 5 – Crystal Violet Kinetics</p> <p>Due Next Class Period:</p> <ol style="list-style-type: none"> 1. Watch the lecture on writing formal lab reports in Sakai (Panopto) and read over PDF materials about formal lab reports (Formal Lab Reports). 2. Type formal lab report on the Crystal Violet lab experiment. 3. Submit Formal Lab Report 1 draft for Peer Review to Sakai (Assignments) 4. Watch lecture on Bromothymol Blue Equilibrium in Sakai (Panopto) 5. Read PDF lab manual for Bromothymol Blue Equilibrium (Sakai Resources) 6. Write necessary info in lab notebook, will be checked at start of class.
Mar 7 th – Mar 11 th , Spring Break	Spring Break - NO CLASS!
Week 8, Mar 11 th – Mar 18 th	<p>Synchronous Lab 6 – Bromothymol Blue Equilibrium</p> <p>Due Next Class Period:</p> <ol style="list-style-type: none"> 1. Submit Lab 6 notebook pages to Sakai (Assignments) 2. Take the quiz on Lab 6 in Sakai (Tests and Quizzes). 3. Watch the lecture for information on the Peer Review Process in Sakai (Panopto) and read over provided PDF materials (Sakai Resources) 4. Complete the assigned peer reviews for Formal Lab Report 1 in Sakai (Assignments)

Week & Class Dates	Activity Planned
Week 9, Mar 22 nd – Mar 25 th	<p>Asynchronous Review peer feedback, submit final report, *absent students must come to lab this week to make up missed in-person lab work!*</p> <p>Due Next Class Period:</p> <ol style="list-style-type: none"> 1. Submit Final Draft of Formal Lab Report 1 in Sakai (Assignments) 2. Watch lecture on the Advanced Acid-Base Titration in Sakai (Panopto) 3. Read PDF lab manual for Advanced Acid-Base Titration (Sakai Resources) 4. Write necessary info in lab notebook, will be checked at start of class.
Week 10, Mar 29 th – Apr 1 st	<p>Synchronous Lab 7 – Advanced Acid-Base Titration, pKa and Molar Mass of Nicotinic Acid</p> <p>Due Next Class Period:</p> <ol style="list-style-type: none"> 1. Submit Lab 7 notebook pages to Sakai (Assignments) 2. Take the quiz on Lab 7 in Sakai (Tests and Quizzes) 3. Watch lecture on the Amino Acid Titration in Sakai (Panopto) 4. Read PDF lab manual for Amino Acid Titration (Sakai Resources) 5. Write necessary info in lab notebook, will be checked at start of class.
Week 11, Apr 5 th – Apr 8 th	<p>Synchronous Lab 8 – Amino Acid Titration</p> <p>Due Next Class Period:</p> <ol style="list-style-type: none"> 1. Write and submit Final Draft of Formal Lab Report 2 (no peer review) to Sakai (Assignments)
Week 12, Apr 12 th – Apr 15 th	<p>Easter Break - NO CLASSES THIS WEEK! (Your other classes may meet Tues and Wed)</p>

Week & Class Dates	Activity Planned
Week 13, Apr 19 th – Apr 22 nd	<p>Asynchronous</p> <p>Lab make ups *absent students must come to lab to make up missed in-person lab work!</p> <p>LATE formal lab report 2</p> <p>Due Next Class Period: Review grades on Sakai, complete any make up lab notebook pages from labs made up today, submit Formal 2 if not submitted on time.</p>
Week 14, Dec 6 th – Dec 10 th	Checkout

Semester calendar schedule of laboratory on the next page



Tentative Semester Calendar of Chem 112 Laboratory, Spring 2022

Month	Tuesday	Wednesday	Thursday	Friday
January 2022	18 Intro, Syllabus, Safety/Glass	19 Intro, Syllabus, Safety/Glass	20 Intro, Syllabus, Safety/Glass	21 Intro, Syllabus, Safety/Glass
	25 Asynch 1 Google Scholar	26 Asynch 1 Google Scholar	27 Asynch 1 Google Scholar	28 Asynch 1 Google Scholar
February 2022	1 Lab 1	2 Lab 1	3 Lab 1	4 Lab 1
	8 Lab 2	9 Lab 2	10 Lab 2	11 Lab 2
	15 Lab 3	16 Lab 3	17 Lab 3	18 Lab 3
	22 Lab 4	23 Lab 4	24 Lab 4	25 Lab 4
March 2022	1 Lab 5	2 Lab 5	3 Lab 5	4 Lab 5
	8 No Classes, Spring Break	9 No Classes, Spring Break	10 No Classes, Spring Break	11 No Classes, Spring Break
	15 Lab 6	16 Lab 6	17 Lab 6	18 Lab 6
	22 Asynch 2/Make Up	23 Asynch 2/Make Up	24 Asynch 2/Make Up	25 Asynch 2/Make Up
	29 Lab 7	30 Lab 7	31 Lab 7	1 Lab 7
April 2022	5 Lab 8	6 Lab 8	7 Lab 8	8 Lab 8
	12 No Class	13 No Class	14 Easter Break No Classes	15 Easter Break No Classes Good Friday
	19 Asynch 3/Make Up	20 Asynch 3/Make Up	21 Asynch 3/Make Up	22 Asynch 3/Make Up
	26 Check out	27 Check out	28 Check out	29 Check out

